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## Key indicators

Single-crystal X-ray study

 $T = 293$  KMean  $\sigma(\text{C}-\text{C}) = 0.002$  Å $R$  factor = 0.046 $wR$  factor = 0.140

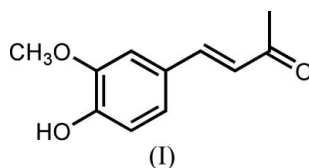
Data-to-parameter ratio = 18.0

For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.**(*E*)-4-(4-Hydroxy-3-methoxyphenyl)but-3-en-2-one**

The title compound,  $\text{C}_{11}\text{H}_{12}\text{O}_3$ , was synthesized from 4-hydroxy-3-methoxybenzaldehyde and acetone. The molecule has a high degree of conjugation throughout the system and intermolecular hydrogen bonds connect adjacent molecules to form one-dimensional chains.

## Comment

$\alpha,\beta$ -Unsaturated ketones are an important class of pharmaceutical intermediates and have extensive application in the synthesis of natural products and drugs by 1,4 addition. As a precursor for further synthesis, we have synthesized the title compound, (I).



All the bond lengths and angles (Table 1) are within normal ranges (Allen *et al.*, 1987) and the structural data confirm the *E* configuration about the  $\text{C}3=\text{C}4$  double bond. Atoms C2, C3, C4 and O1 constitute a well defined plane and the benzene ring plane is inclined at  $5.34(1)^\circ$  to this plane (Fig. 1). A weak intermolecular  $\text{O}2-\text{H}2\cdots\text{O}1$  hydrogen bond links adjacent molecules, forming one-dimensional chains (Fig. 2).

## Experimental

4-Hydroxy-3-methoxybenzaldehyde (3.04 g, 20 mmol) was dissolved in 25 ml of acetone and 12 ml of dilute aqueous NaOH solution (10%) was added to the acetone solution with stirring. The mixture was allowed to stand overnight at room temperature and then the mixture was acidified with dilute aqueous HCl to give (*E*)-4-(4-hydroxy-3-methoxyphenyl)but-3-en-2-one as a yellow solid (yield 73%). The resultant precipitate was filtered off, washed with water and recrystallized from ethanol and dichloromethane (2:1), in air over a period of four days. After about three-quarters of the original solvent had evaporated, large colourless prisms of (I) were obtained (yield 61%).

## Crystal data

$\text{C}_{11}\text{H}_{12}\text{O}_3$   
 $M_r = 192.21$   
Monoclinic,  $P2_1/c$   
 $a = 9.602(5)$  Å  
 $b = 7.780(4)$  Å  
 $c = 13.478(7)$  Å  
 $\beta = 97.466(8)^\circ$   
 $V = 998.2(9)$  Å<sup>3</sup>  
 $Z = 4$

$D_x = 1.279$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation  
Cell parameters from 2995  
reflections  
 $\theta = 2.6-27.9^\circ$   
 $\mu = 0.09$  mm<sup>-1</sup>  
 $T = 293(2)$  K  
Prism, colourless  
 $0.36 \times 0.28 \times 0.22$  mm

Data collection

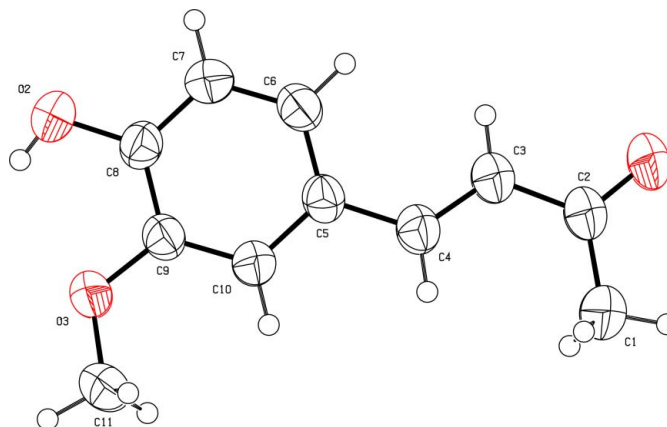
Bruker SMART CCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: none  
 8188 measured reflections  
 2355 independent reflections

1816 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.020$   
 $\theta_{max} = 28.4^\circ$   
 $h = -12 \rightarrow 12$   
 $k = -10 \rightarrow 10$   
 $l = -17 \rightarrow 17$

Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.046$   
 $wR(F^2) = 0.140$   
 $S = 1.04$   
 2355 reflections  
 131 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.07P)^2 + 0.1629P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{max} = 0.001$   
 $\Delta\rho_{max} = 0.20 \text{ e } \text{Å}^{-3}$   
 $\Delta\rho_{min} = -0.19 \text{ e } \text{Å}^{-3}$   
 Extinction correction: *SHELXL97*  
 Extinction coefficient: 0.024 (5)



**Figure 1**  
 The structure of the title compound (1), showing 50% probability displacement ellipsoids and the atom-numbering scheme.

**Table 1**

Selected geometric parameters (Å, °).

|           |             |           |             |
|-----------|-------------|-----------|-------------|
| O2—C8     | 1.3556 (18) | C10—C9    | 1.3785 (19) |
| C4—C3     | 1.336 (2)   | C8—C7     | 1.380 (2)   |
| C4—C5     | 1.459 (2)   | C8—C9     | 1.394 (2)   |
| C5—C6     | 1.388 (2)   | C6—C7     | 1.379 (2)   |
| C5—C10    | 1.400 (2)   | O1—C2     | 1.2243 (18) |
| O3—C9     | 1.3686 (17) | C3—C2     | 1.454 (2)   |
| O3—C11    | 1.413 (2)   | C2—C1     | 1.490 (2)   |
| C3—C4—C5  | 127.19 (15) | O3—C9—C10 | 125.48 (13) |
| C6—C5—C10 | 118.55 (12) | O3—C9—C8  | 114.63 (12) |
| C6—C5—C4  | 122.60 (13) | C10—C9—C8 | 119.89 (13) |
| C10—C5—C4 | 118.85 (13) | C7—C6—C5  | 120.63 (14) |
| C9—O3—C11 | 117.63 (12) | C4—C3—C2  | 124.57 (15) |
| C9—C10—C5 | 120.79 (13) | O1—C2—C3  | 119.85 (15) |
| O2—C8—C7  | 118.81 (13) | O1—C2—C1  | 119.74 (14) |
| O2—C8—C9  | 121.70 (13) | C3—C2—C1  | 120.40 (14) |
| C7—C8—C9  | 119.48 (13) | C6—C7—C8  | 120.64 (14) |

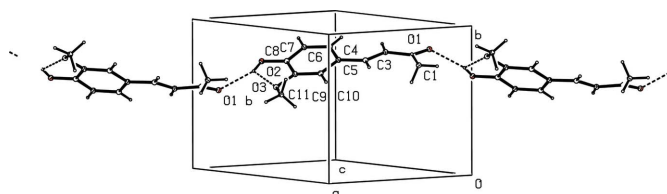
**Table 2**

Hydrogen-bond geometry (Å, °).

| $D-H \cdots A$                 | $D-H$ | $H \cdots A$ | $D \cdots A$ | $D-H \cdots A$ |
|--------------------------------|-------|--------------|--------------|----------------|
| O2—H2 $\cdots$ O1 <sup>i</sup> | 0.82  | 2.08         | 2.772 (2)    | 141            |

Symmetry code: (i)  $x + 1, -y + \frac{1}{2}, z - \frac{1}{2}$ .

H atoms were positioned geometrically [C—H distances of 0.93 ( $C_{sp^2}$ —H) and 0.96 Å (methyl), and O—H = 0.82 Å].  $U_{iso}(H)$  values were set equal to  $xU_{eq}(\text{carrier atom})$ , where  $x = 1.2$  for CH and 1.5 for O and methyl C.



**Figure 2**  
 A view of part of the crystal packing. Dashed lines indicate hydrogen bonds.

Data collection: *SMART* (Bruker, 2002); cell refinement: *SAINT* (Bruker, 2002); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Sheldrick, 2000); software used to prepare material for publication: *SHELXTL*.

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References

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